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Larson, C. William; Harper, Jessica; Presilla-Marquez, J.D. (Schafer Corp.), "Matrix Isolation of Boron and Carbon Vapor. Control of Cluster Formation During Preparation and Annealing"

**10<sup>th</sup> Internat'l IUPAC Conference on High Temperature Materials Chemistry (Statement A)**  
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# MATRIX ISOLATION OF BORON AND CARBON VAPOR. CONTROL OF CLUSTER FORMATION DURING PREPARATION AND ANNEALING

C. W. Larson, J. Harper and J. D. Presilla-Márquez<sup>a</sup>

Air Force Research Laboratory

Edwards AFB, CA 93524-7680

E-mail: carl.larson@ple.af.mil

## Abstract

The formation and recombination of small boron/carbon clusters were studied by quantitative Fourier transform infrared (FTIR) spectroscopy. Samples were prepared by evaporation of powder mixtures from a resistively heated oven (3000 K) and isolated at 10 K in solid argon.  $C_3$  and  $BC_2$  disappeared entirely over the course of several annealings. Yields of all the larger clusters ( $B_JC_{n-J}$ ,  $n \geq 4$ ,  $J = 0, 1, 2$ ) increased during annealing. Clusters with  $n \geq 4$  are linear, with the boron atoms of the  $J = 1$  and 2 clusters capping the ends of carbon chains;  $B_2C_{n-2}$  clusters are inert. During annealing,  $C_n$  and  $BC_{n-1}$  clusters grow by adding a mobile atom, dimer, or trimer to the carbon end of the chain. Dramatic growth of  $B_2C_2$  during the first annealing is consistent with the presence of B atoms and/or BC molecules in the initial matrices. (do not react to form larger clusters w/in matrix)

## Introduction

Under the auspices of the U. S. Air Force High Energy Density Matter (HEDM) Program [1], we have pursued the goal of production of a cryogenic matrix with an atom density of  $\sim 1 \times 10^{21}$  atoms  $cm^{-3}$ , which amounts to  $\sim 5$  mole percent atoms in an inert solid, or an equivalent atom gas pressure of  $\sim 60$  atm at 298 K. This and our companion paper [2] describe qualitative and quantitative analysis of  $B_JC_{n-J}$  clusters in prototypical HEDM. Previously, the first conclusive identifications of small boron clusters,  $BC_2$ ,  $B_2C$ ,  $BC_3$ , and  $B_2C_2$  were reported [3-5]. Observed FTIR fundamentals and isotope shifts were matched with predictions of *ab-initio* theory in those studies.

Verhaegen, Stafford, and Drowart [6] published the first experimental evidence of  $B_JC_{n-J}$  35-years ago. Mass spectrometry was used to measure boron and boron carbide vaporization from graphite Knudsen cells heated by electron bombardment to  $\sim 2500$  K. The vapor pressure of boron at 2200 K was  $\sim 0.008$  torr, 94% atoms, 5%  $BC_2$ , 1%  $B_2C$ , and 0.07% BC. At the same temperature, boron carbide had about one-half the vapor pressure with 96% atoms, 4%  $BC_2$ , 0.1%  $B_2C$ . Distributions of anions and cations of  $B_xC_y$  with up to 17-atoms have been measured in plumes of laser vaporized boron carbide targets by Becker and Dietze [7]. Their mass spectrometric analysis showed that  $B_2C^-$ ,  $B_2C^+$ ,  $BC_2^-$ , and  $BC_2^+$  were major components of the plume. Matrix isolated  $BC_2$  has been produced by oven and laser vaporization of carbon/boron mixtures and targets and measured by FTIR [3,8] and ESR [9,10] spectroscopy. Most recently, Wyss, Grutter and Maier [11] isolated  $BC_2^-$  and  $BC^-$  in neon matrices by sputtering a boron carbide target with  $Cs^+$  and trapping the mass-selected anions in neon matrices. Neutrals were produced by electron photodetachment with uv radiation. Electronic transitions of  $BC_2$  and BC and their anions were studied. The FTIR

<sup>a</sup> Present address: Schafer Corporation, 26565 W. Agoura Rd., Calabasas, California 93102.

spectrum of the previously identified [3]  $\nu_2(a_1)$  fundamental of  $BC_2$  was also observed. The infrared spectrum of BC has never been measured, but a CCSD(T)/cc-pVTZ calculation with quartic force field predicted [12] its vibrational fundamental at  $1148\text{ cm}^{-1}$  with intensity of  $43\text{ km mol}^{-1}$ . Of the pure boron clusters, only  $B_3$  has been experimentally observed. Li, Van Zee and Weltner [13] identified all four isotopomers of  $B_3$  in argon matrices by FTIR spectroscopy and found that experimental frequencies matched *ab-initio* frequencies of its cyclic geometry. Although *ab-initio* studies of  $B_n$ ,  $n \geq 4$ , have been carried out, only geometries and energies have been reported [14,15]. Unlike  $C_n$ ,  $B_n$  for  $n \geq 5$  are predicted to have stable 3-dimensional geometries.

Carbon clusters were recently reviewed by Van Orden and Saykally [16]. Cyclic  $C_6$  [17, 18], cyclic  $C_8$  [19, 20], linear  $C_{11}$  [21], have recently been identified by their FTIR isotopomer spectroscopy. Most recently, electronic spectra of linear  $C_{17}$ ,  $C_{19}$ , and  $C_{21}$  [22] and cyclic  $C_{10}$  and cyclic  $C_{12}$  [23] isolated in neon matrices were reported, and Raman spectra of linear  $C_{16}$ ,  $C_{18}$ , and  $C_{20}$  were reported [24].

Figure 1 shows a Pascal's triangle representation of the boron/carbon system. In the limit of statistical recombination of well mixed atoms, the relative yields of  $B_JC_{n-J}$  clusters of given  $n$  is given by

$$\rho(B_JC_{n-J})/\rho(C_n) = n!/J!(n-J)! [B/C]^J, \quad (1)$$

where  $[B/C]$  is the molar ratio of boron to carbon, and  $\rho(i)$  is the column density of species  $i$ .

## Experimental

Mixtures of boron and carbon powders were evaporated with a resistively heated oven to temperatures  $\sim 3000\text{ K}$ , and vapors were co-condensed with  $\sim 1000:1$  excess argon at  $10\text{ K}$  to form  $\sim 60\text{ }\mu$  thick matrices in about one-hour. Beer's Law enabled quantitative analysis of the various species,  $\rho(i) = 2.303 A_i(\text{experiment})/I_i(\text{theory})$ , where  $\rho(i)$  is the average column density, which is the average value of the product of the species density and matrix thickness.  $A_i(\text{experiment})$  is the integrated intensity of the infrared absorbance of species  $i$ , equal to  $-\log_{10} \tau_i(\nu)$ , where  $\tau_i(\nu)$  is the transmittance at absorption frequency  $\nu$ .  $I_i(\text{theory})$  is the infrared absorption intensity of species  $i$ , which by necessity must be obtained by *ab-initio* calculation because experimental values are not available at this time. Fig. 1 indicates values of  $I_i(\text{theory})$  calculated by DFT/cc-pVDZ [25-28].

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## Results and Discussion

Annealing of pure carbon matrices led to disappearance of atoms, dimers and trimers from the initial deposits. The identified products of recombination included cyclic  $C_6$  and cyclic  $C_8$ , and linear  $C_n$  molecules,  $4 \leq n \leq 13$ . After annealing to constant composition, the total measurable carbon was found to be as much as twice the initial measurable carbon, indicating that about 50 % of the initially deposited carbon was isolated as atoms and dimers. The cyclic molecules, the dominant condensation products, amounted to as much as  $\sim 60\%$  of total carbon in the fully annealed matrices.

In matrices containing boron, linear  $C_3$  and cyclic  $BC_2$ ,  $B_2C$ , and  $B_3$ , constituted about 80% of the total observable boron and carbon in the initially deposited matrix. The observation of approximately statistical distributions [Equation (1)] in clusters with  $n = 3, 4$ , and 5 in the initially deposited matrices implies that these smaller clusters form by random condensation of well-mixed atoms, uninfluenced by their relative energies, the energies of their precursors, or preferential kinetics pathways that could otherwise distort the statistics.

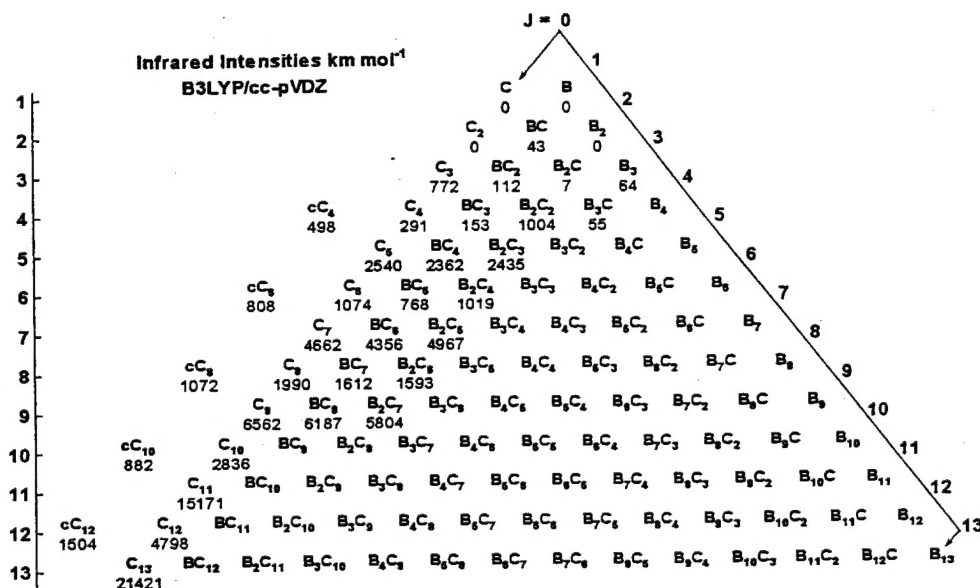


Figure 1. Representation of the  $B_nC_{n-J}$  ( $n = 1 - 13$ ,  $J = 0 - n$ ) system by Pascal's triangle. Numbers below molecular formulas denote the *ab-initio* intensities of the most intense infrared stretching fundamentals, Refs. [25] – [28]. Cyclic carbon clusters are removed from the triangle to the left.

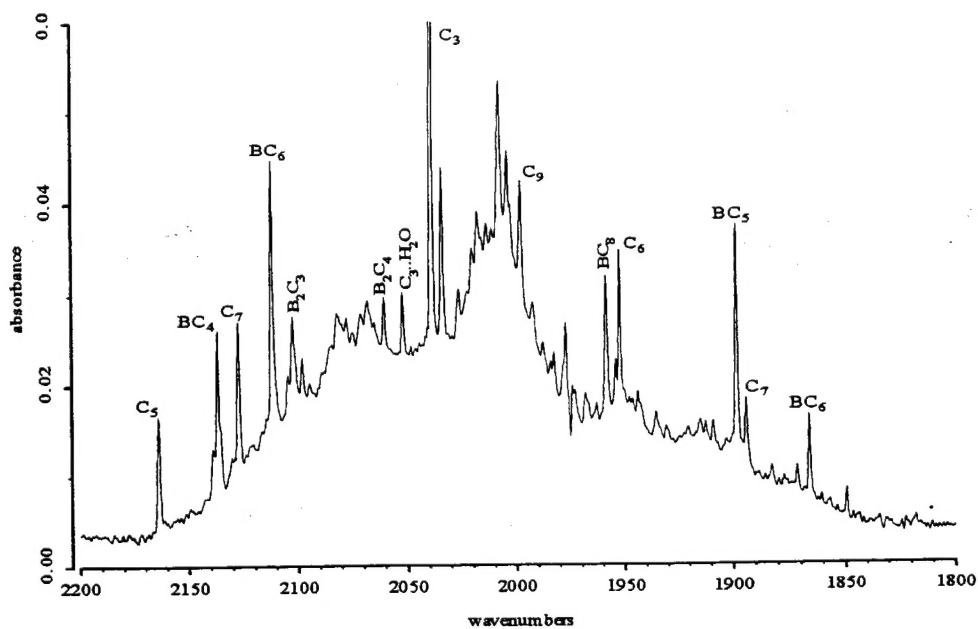


Figure 2. Survey spectrum of matrix containing carbon and boron at natural abundance with  $[B/C] \sim 1/3$  after three annealings. All of the peaks indicated grow upon annealing except  $C_3$ . Fundamentals of  $BC_{n-1}$  for  $n = 5, 6, 7$ , and  $9$  are similarly red-shifted from fundamentals of linear  $C_n$  and their experimental absorbances are all slightly greater in this matrix. Two fundamentals of  $BC_6$  are observed at  $2112$  and  $1866 \text{ cm}^{-1}$ , red-shifted from the two fundamentals of linear  $C_7$ .

$B_2C_{n-2}$  clusters, once formed, were inert to further condensation upon annealing. Thus, energy loss by recombination during preparation of high energy density matrices (HEDM) is minimized in matrices with large B/C ratios. For example, when B/C  $\sim 2$ , any recombination that does occur during co-deposition would tend to produce  $B_2C$ , a high-energy molecule that is inert in the cryogenic environment. In pure boron HEDM, the cyclic  $B_3$  molecule may be similarly inert, which would suggest that energy loss by recombination will also stop at a  $B_3$  "island of stability." Continuing experimental work to produce five-mole percent pure boron HEDM is concentrating on development of a resistively heated high-flux boron atom source, the "boron cannon", and production of higher density matrices.

## References

1. M. R. Berman, Editor, *Proceedings of the High Energy Density Matter (HEDM) Contractors' Conference* (Monterey, California, May 1998), Air Force Office of Scientific Research, Bolling AFB, DC 20332-8050, and annual proceedings dating from 1987.
2. C. W. Larson, J. Harper, J. D. Presilla-Márquez, Poster Paper, **This Volume**.
3. J. D. Presilla-Márquez, C. W. Larson, P. G. Carrick, and C. M. L. Rittby, *J. Chem. Phys.* **105**, 3398 (1997).
4. J. D. Presilla-Márquez, P. G. Carrick, and C. W. Larson, *J. Chem. Phys.* **110**, 5702 (1999).
5. C. W. Larson and J. D. Presilla-Márquez, *J. Chem. Phys.* **111**, 1988 (1999).
6. G. Verhaegen, F. E. Stafford, and J. Drowart, *J. Chem. Phys.* **40**, 1622 (1964).
7. S. Becker, and H. J. Dietze, *Intl. J. Mass Spectrometry and Ion Proc.* **82**, 287 (1988).
8. J. M. L. Martin, P. R. Taylor, J. T. Yustein, T. R. Burkholder, and L. Andrews, *J. Chem. Phys.* **99**, 12 (1993).
9. W. C. Easley and W. Weltner, Jr., *J. Chem. Phys.* **52**, 1489 (1970).
10. L. B. Knight, Jr., S. Cobranchi, E. Earl, and A. J. McKinley, *J. Chem. Phys.* **104**, 4927 (1996).
11. M. Wyss, M. Grutter, and J. P. Maier, *J. Phys. Chem. A* **102**, 9106 (1998).
12. J. M. L. Martin and P. R. Taylor, *J. Chem. Phys.* **100**, 9002 (1994).
13. S. Li, R. J. Van Zee, and W. Weltner, Jr., *Chem. Phys. Lett.* **262**, 298 (1996).
14. J. Niu, B. K. Rao and P. Jena, *J. Chem. Phys.* **107** 132 (1997).
15. I. Boustani, *Chem. Phys. Lett.* **233** 273 (1995), 240 135 (1995).
16. A. Van Orden, and R. J. Saykally, *Chem. Rev.* **98**, 2313 (1998).
17. J. D. Presilla-Márquez, J. A. Sheehy, J. D. Mills, P. G. Carrick, and C. W. Larson, *Chem. Phys. Lett.* **274**, 439 (1997).
18. S. L. Wang, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **107**, 6032 (1997).
19. J. D. Presilla-Márquez, J. Harper, J. A. Sheehy, P. G. Carrick, and C. W. Larson, *Chem. Phys. Lett.* **300**, 719 (1999).
20. S. L. Wang, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **107**, 7025 (1997), **112**, 1457 (2000), and X. D. Ding, et al., *J. Chem. Phys.* **112**, 5113 (2000).
21. L. Lapinski and M. Vala, *Chem. Phys. Lett.* **300**, 195 (1999).
22. M. Wyss, M. Grutter, and J. P. Maier, *Chem. Phys. Lett.* **304**, 35 (1999).
23. M. Grutter, M. Wyss, E. Riaplov, and J. P. Maier, *J. Chem. Phys.* **111**, 7397 (1999).
24. A. K. Ott, G. A. Rechtsteiner, C. Felix, O. Hampe, M. F. Jarrold, R. P. Van Duyne, and K. Raghavachari, *J. Chem. Phys.* **109**, 9652 (1998).
25. C. M. L. Rittby, J. A. Sheehy, and M. Vala, M. Zerner (personal communications).
26. J. M. L. Martin, J. P. François, and R. Gijbels, *J. Chem. Phys.* **90**, 6469 (1989).
27. J. M. L. Martin, J. El-Yazal, and J. P. François, *Chem. Phys. Lett.* **242**, 570 (1995).
28. J. M. L. Martin and P. R. Taylor, *J. Chem. Phys.* **100**, 9002 (1994).

# MATRIX ISOLATION OF BORON AND CARBON VAPOR.

C. W. Larson, J. Harper and J. D. Presilla-Márquez

*Air Force Research Laboratory*

*Edwards AFB, CA 93524-7680*

*E-mail: carl.larson@ple.af.mil*

## Abstract

This report supplements the companion paper published in this volume [1]. Details about the FTIR quantitative analysis of the disappearance and formation of clusters during annealing of solid argon matrices containing boron and carbon species are presented.

## Introduction

The companion paper [1] summarized the literature of small boron clusters, about  $B_JC_{n-J}$  ( $n = 2$  to 10,  $J = 0, 1, 2, 3$ ) and described the formation of clusters during preparation and annealing of solid argon. This report presents additional details.

## Experimental

A precision matched pair of matrices was prepared as described in Ref. [1] with two different boron isotopic compositions and natural abundance carbon. Matrix (a) contained natural abundance boron,  $^{11}B/^{10}B = 4.0$ , and matrix (b) contained boron with an inverted isotope ratio,  $^{11}B/^{10}B = 0.37$ . Each matrix had a  $[C/B]$  molar ratio of  $\sim 3.0$ , a thickness of  $\sim 60$  m, and very similar absolute amounts of each element. Each matrix was annealed repeatedly according to the same annealing protocol given in the caption of Figure 1.

## Results and Discussion

Figure 1 shows the annealing behaviors of the FTIR bands belonging to  $BC_2$ ,  $BC_3$  and  $B_2C_2$  for each of the matrices. Cyclic  $BC_3$  disappears entirely when one of its carbon atoms is attacked by a radical species. One of the B-C bonds is broken, and the  $BC_3$  ring opens. The dramatic appearance of  $B_2C_2$  during the first annealing is consistent with a mechanism involving recombination of BC, whose density lies below the limit of detection, or condensation of B atoms with  $BC_2$ . → with

Figure 2 shows the annealing behaviors of the  $^{11}B_J^{12}C_{n-1}$  clusters,  $n = 3$  to 9 and  $J = 0, 1, 2$  that were measured in matrix (a). Loss of clusters during the sixth annealing by matrix sublimation is generally greatest for  $C_n$  and smallest for  $B_2C_{n-2}$ , which is consistent with a stratified matrix with a higher  $[C/B]$  ratio near the subliming surface. Apparently, boron evaporates more rapidly than carbon during the early stages of oven evaporation, resulting in boron-rich layers near the substrate.

Column densities of the trimer clusters in the initially deposited matrices are in the ratio  $\rho(C_3):\rho(BC_2):\rho(B_2C):\rho(B_3) \approx 1.0:1.5:0.5: < 0.05$ , where only the upper limit for  $\rho(B_3)$  may be determined because it was not observed. This is consistent with formation of the trimers by statistical condensation from a matrix with  $[B/C] \sim 1/3$ , which would produce a  $1.0:1.0:0.33:0.05$  distribution,  $\rho(B_JC_{n-J})/\rho(C_n) = n!/J!(n-J)! [B/C]^J$ , Eq. (1), Ref. [1].

Agreement between distributions implies trimers form by random condensation of well-mixed atoms, uninfluenced by the relative energies of the trimers, the energies of their precursors, or preferential kinetics pathways that could otherwise distort the statistics. Linear  $C_3$  and cyclic  $BC_2$ , disappeared entirely when the matrices were repeatedly annealed to



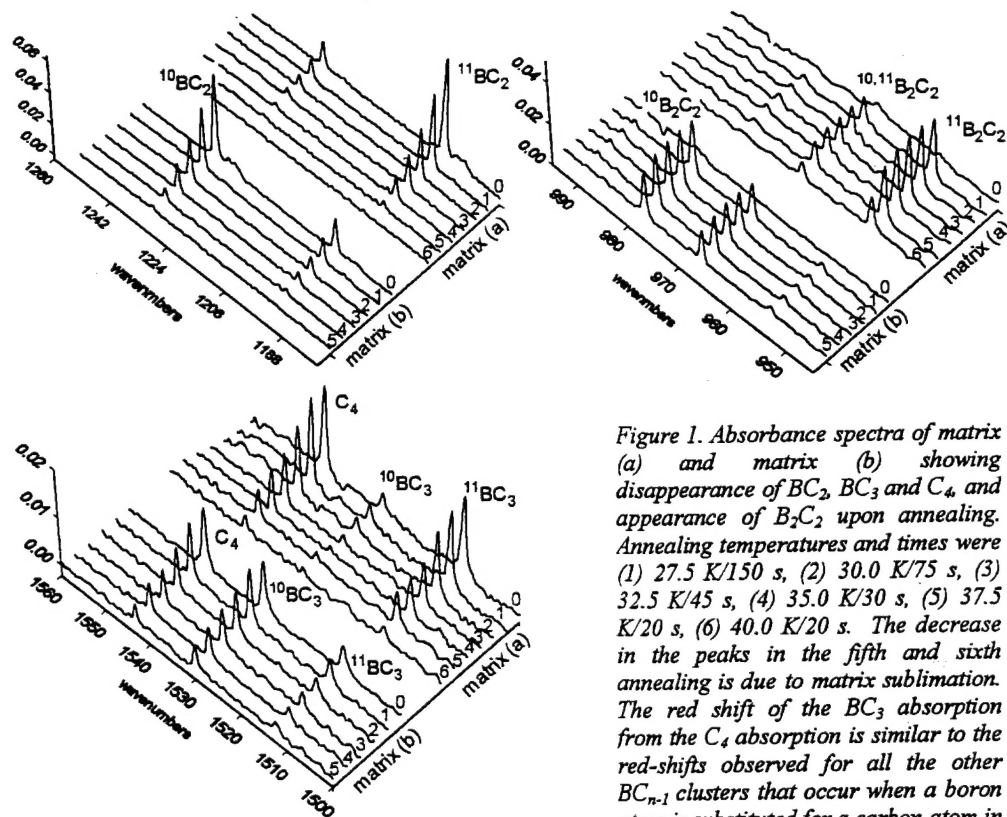


Figure 1. Absorbance spectra of matrix (a) and matrix (b) showing disappearance of  $BC_2$ ,  $BC_3$  and  $C_4$  and appearance of  $B_2C_2$  upon annealing. Annealing temperatures and times were (1) 27.5 K/150 s, (2) 30.0 K/75 s, (3) 32.5 K/45 s, (4) 35.0 K/30 s, (5) 37.5 K/20 s, (6) 40.0 K/20 s. The decrease in the peaks in the fifth and sixth annealing is due to matrix sublimation. The red shift of the  $BC_3$  absorption from the  $C_4$  absorption is similar to the red-shifts observed for all the other  $BC_{n-1}$  clusters that occur when a boron atom is substituted for a carbon atom in a  $C_n$  cluster.

temperatures between 25 K and 35 K, but cyclic  $B_2C$  was inert. Linear  $C_4$  and  $BC_3$  (BCCC) disappeared more slowly, and linear  $B_2C_2$  (BCCB) grew to  $\sim 95\%$  of its final value during the first annealing. Once formed,  $B_2C_2$ , like  $B_2C$ , was also inert to further reaction.

Figure 3 shows the absolute column densities of the analyzed clusters that <sup>(was)</sup> obtained by transformation of the Fig. 2 data with Beer's law and theoretical intensities shown in Fig. 1, Ref [1]. The  $BC$  molecule was not observed, but its upper limit column density was established to be somewhat larger than that of  $B_2C_2$ .

The sources of  $B_2C_2$  are from condensation of atom plus trimer ( $B + BC_2$  but not  $C + B_2C$ ) or dimer + dimer ( $BC + BC$  but not  $B_2 + C_2$ ). Although  $BC$  was not observed, the upper limit of  $\rho(BC)$  is larger than  $\rho(B_2C_2)$  so that  $BC$  cannot be ruled out as a source of  $B_2C_2$ . The growth of  $B_2C_2$  is consistent with the presence of  $BC$  and/or  $B$  in the originally deposited matrix in an amount at least as great as the growth of  $B_2C_2$ .

Figures 1 and 2 show that  $C_4$  and  $BC_3$  do not change during the first annealing but begin to disappear in subsequent annealings. This indicates that a source for these tetramers, atom + trimer and/or dimer + dimer, operates during the first annealing.

Figure 2 shows that linear  $C_5$ ,  $BC_4$  (BCCCC) and  $B_2C_3$  (BCCCB) and larger linear clusters all grew upon annealing. The sources of  $B_2C_3$  are dimer + trimer ( $BC + BC_2$  but not  $B_2 + C_3$ ) and atom + tetramer ( $B + BC_3$  but not  $C + B_2C_2$ ). Since  $\rho(BC_2) \sim 5\rho(BC_3)$  in the initially deposited matrix, the  $BC + BC_2$  source is dominant. Growth of  $B_2C_3$  is consistent with the presence of  $BC$  in the initially deposited matrix in an amount at least as great as the amount by which  $B_2C_3$  grows.



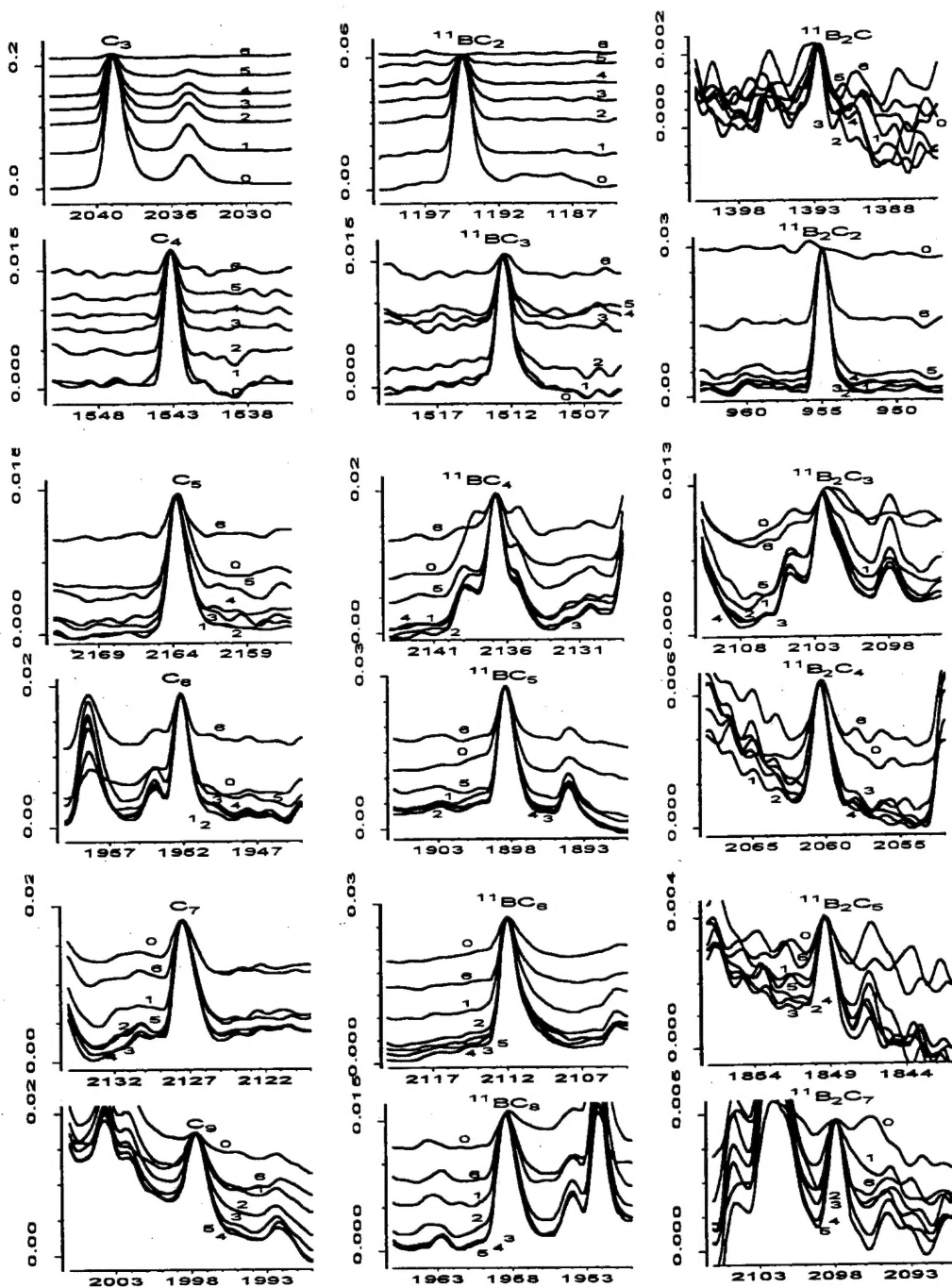
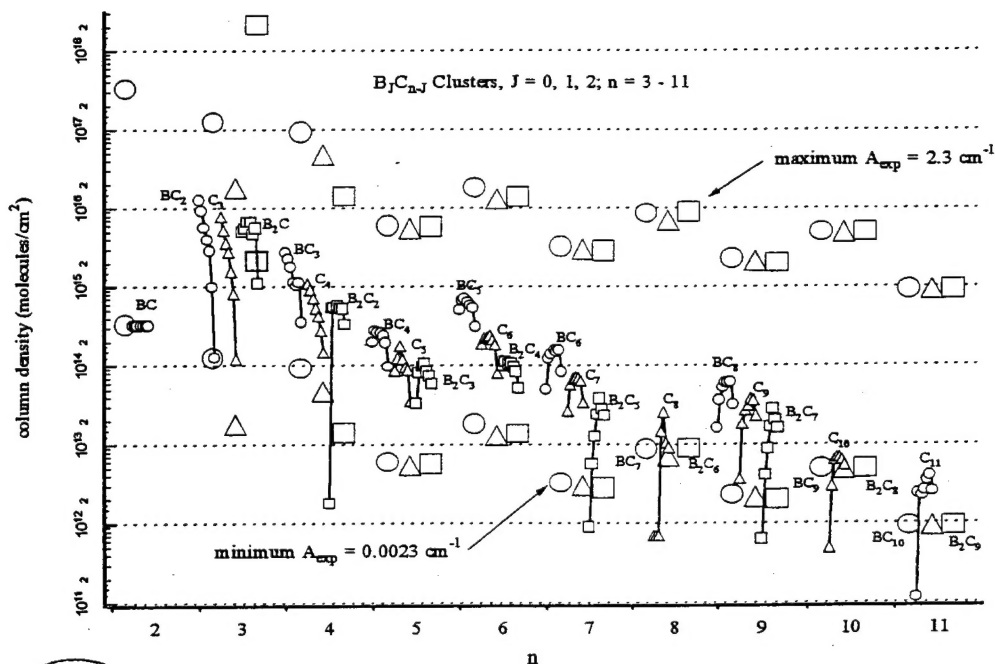


Figure 1. Annealing behaviors of  $^{11}\text{B}_j^{12}\text{C}_{n-j}$  species in matrix (a), which contained natural abundance carbon and boron,  $^{11}\text{B}/^{10}\text{B}=80/20$ ,  $^{12}\text{C}/^{13}\text{C}=99/1$ , with  $[\text{C}/\text{B}] \sim 3$ . Spectra labeled '0' were obtained from the originally deposited matrix, and spectra labeled '1' to '6' were obtained after successive annealings. During the 6th annealing (at 40 K for 20 seconds), almost half the matrix evaporates.



3. Figure 2. Distributions of  $B_J C_{n-J}$  clusters,  $J = 0, 1, 2$ ;  $n = 3 - 11$ . Circles, triangles and squares represent  $BC_{n-1}$ ,  $C_n$  and  $BC_{n-2}$  clusters, respectively. Large symbols denote upper and lower limits of measurement, based on a nominal minimum measurable absorbance of  $0.0023 \text{ cm}^{-1}$ , and a maximum absorbance for linearity of Beer's law of  $2.3 \text{ cm}^{-1}$  (1% transmittance). Small symbols denote measured quantities in the initial matrix, and in six annealed matrices. Some of the larger clusters ( $n = 8, 10, 11$ ) have not been identified,  $BC_7$ ,  $B_2C_6$ ,  $BC_9$ ,  $B_2C_8$ ,  $BC_{10}$ ,  $B_2C_9$ .

Growth of  $BC_4$  occurs primarily by  $BC + C_3$  rather than  $B + C_4$  or  $C + BC_3$  because  $\rho(C_3) \sim 10\rho(C_4)$  and  $\rho(C_3) \sim 2\rho(BC_3)$ . Growth of  $C_5$  occurs by  $C + C_4$  and  $C_2 + C_3$ , which establishes the presence of  $C$  and/or  $C_2$  in the original matrix in an amount at least as great as  $C_5$  growth.

## Conclusions

Disappearance of triangular  $BC_2$  requires breaking of one of its B-C bonds when one of its carbon atoms is attacked. The major reorganization of electronic energy involved in opening the ring appears to occur with little ( $< \sim 3 \text{ kcal mol}^{-1}$ ) or no energy barrier, which makes this small molecule a candidate for an interesting *ab-initio* study of unusual reactivity at low temperature. Annealing kinetics of disappearance of  $C_3$  and  $BC_2$ , and of appearance of  $B_2C$ ,  $C_4$ ,  $BC_3$ ,  $B_2C_2$ ,  $C_5$ ,  $BC_4$ , and  $B_2C_3$  unequivocally establishes the presence of atoms and dimers in the originally deposited matrix. About 80% or more of the initially deposited HEDM existed as atoms, dimers and trimers. Molecules with two boron atoms are immune from radical attack and condensation during annealing.

## Reference

1. C. W. Larson, J. Harper, J. D. Presilla-Márquez, Poster Paper, This Volume.